

# Experimental report

10/10/2019

**Proposal:** 5-24-613

**Council:** 4/2018

**Title:** Reaction pathways of hydrogenation-dehydrogenation of InPd<sub>3</sub> and FePd<sub>3</sub> by in situ neutron diffraction

**Research area:** Chemistry

**This proposal is a new proposal**

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**Samples:** FePd<sub>3</sub>  
InPd<sub>3</sub>

Instrument	Requested days	Allocated days	From	To
D20	3	3	05/10/2018	08/10/2018

## Abstract:

The hydrogen-induced atomic rearrangements in intermetallic compounds MPd<sub>3</sub> (M = In, Fe) will be studied by in situ neutron powder diffraction. A user-supplied gas-pressure cell with laser heating installed on D20 will provide the means to collect diffraction patterns as a function of temperature (up to 600 K) and deuterium gas pressure (up to 100 bar) with a time resolution of two minutes. This will allow to identify intermediates including deuterium positions in the crystal structures and provide an understanding of the underlying principle of hydrogen-induced rearrangement in intermetallics. For InPd<sub>3</sub>, the series of superstructures of the cubic-close packing (ccp) TiAl<sub>3</sub> type - ZrAl<sub>3</sub> type - PbTi<sub>2</sub>Pd<sub>9</sub> type - AuCu<sub>3</sub> type is to be monitored upon hydrogen uptake. For FePd<sub>3</sub>, the effect of atomic order (Fe/Pd, AuCu<sub>3</sub> type and Cu type) on hydrogenation and hydrogen-induced ordering will be elucidated. These reactions are of importance in hydrogen embrittlement of materials and in catalytic processes.

## Reaction pathways of hydrogenation-dehydrogenation of $\text{InPd}_3$ and $\text{FePd}_3$ by *in situ* neutron diffraction (experiment 5-24-613)

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$\text{InPd}_3$  crystallizes in ordered superstructures of the cubic close packing (ccp), the  $\text{ZrAl}_3$  and the  $\text{TiAl}_3$  type. Upon hydrogen uptake, a rearrangement to another ccp superstructure, the  $\text{AuCu}_3$  type, occurs with hydrogen atoms occupying octahedral  $[\text{Pd}_6]$  voids [1]. The hydrogenation and dehydrogenation was followed *in situ* in this experiment on a  $\text{ZrAl}_3$  type sample using deuterium in a sapphire single-crystal cell [2]. When increasing deuterium pressure at 400 K to 5.0 MPa, a new phase is formed,  $\text{InPd}_3\text{D}_{<0.7}$  with a deuterium filled  $\text{ZrAl}_3$  type structure. Raising temperature to 500 K leads to the cubic deuterides  $\text{InPd}_3\text{D}_{\approx 0.8}$  with a cubic antiperovskite type structure [1]. Heating in vacuum leads to a decrease in deuterium content of the cubic deuterides, and upon complete deuterium release  $\text{ZrAl}_3$  type  $\text{InPd}_3$  is restored (Fig. 1). Due to the sluggish reaction of the  $\text{ZrAl}_3$  type sample and time constraints ( $\text{FePd}_3$  samples see below) the  $\text{TiAl}_3$  type sample could not be measured as foreseen in the proposal text.

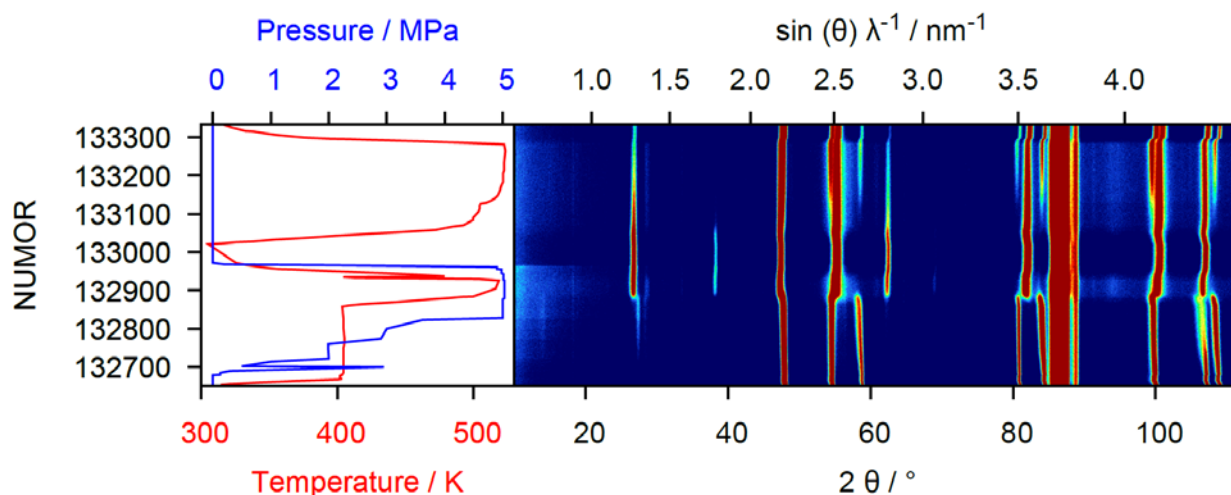


Figure 1: *In situ* neutron powder diffraction data of the deuteration of  $\text{InPd}_3$  taken on diffractometer D20 at  $\lambda = 1.86796(3)$  Å in a single crystal sapphire cell [2] under various deuterium pressures and temperature conditions. Intensities are in false colors, each measurement (NUMOR) with two minutes data collection time.

To study the influence of structural order in  $\text{FePd}_3$  on hydrogenation properties, *in situ* neutron powder diffraction on an ordered and a disordered sample using deuterium in a sapphire single-crystal cell [2] was performed. Deuterium gas pressure was slowly increased up to 8.0 MPa, before raising the temperature to about 550 K (Fig. 2).  $\text{FePd}_3$  with ordered and disordered structure showed less deuterium uptake than expected from previously executed laboratory X-ray diffraction experiments. For the disordered compound the unit cell increases only by 0.05% ( $a_{\text{before}} = 3.8512(4)$  Å,  $a_{\text{after}} = 3.8518(2)$  Å) and the site occupancy factor (SOF) of deuterium is refined to zero (Wyckoff position 4b), i. e. it does not take up a significant amount of deuterium. Furthermore, no ordering of

the metal atoms is observed. The unit cell volume of ordered FePd<sub>3</sub> expanded by 0.2% ( $a(\text{FePd}_3) = 3.85158(9) \text{ \AA}$ ,  $a(\text{FePd}_3\text{D}_{0.03(1)}) = 3.85416(2) \text{ \AA}$ ) during deuteration. The deuterium atoms occupy [Pd<sub>6</sub>] octahedral voids as determined by difference Fourier analysis. During the *in situ* experiment (Fig. 2) the superstructure reflections of FePd<sub>3</sub> and its deuteride are broadened due to very small magnetic domains of the ferromagnetic compound analogous to MnPd<sub>3</sub> [3]. The width of these reflections decrease when reaching the Curie temperature ( $T_C(\text{FePd}_3) = 499.1 \text{ K}$  [4]). Additionally, the intensities of the superstructure reflection decrease and  $\beta\text{-PdD}_x$  is formed. Both effects are reversible during the cooling process.

Complete serial refinements based on the *in situ* diffraction data for both FePd<sub>3</sub> and InPd<sub>3</sub> samples are in progress.

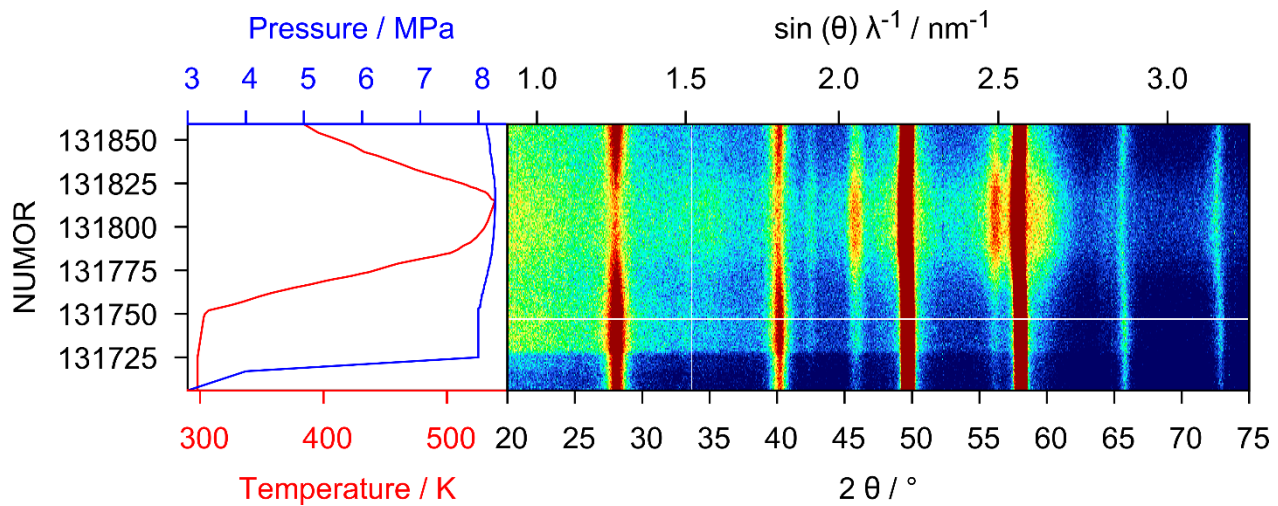


Figure 2: *In situ* neutron powder diffraction data of the deuteration of ordered FePd<sub>3</sub> taken on diffractometer D20 at  $\lambda = 1.86796(3) \text{ \AA}$  in a single crystal sapphire cell [2] under various deuterium pressures and temperature conditions. Intensities are in false colors, each measurement (NUMOR) with two minutes data collection time.

## Literature

- [1] H. Kohlmann, A. V. Skripov, A. V. Soloninin, T. J. Udovic, *J. Solid State Chem.* **2010**, *183*, 2461-2465
- [2] A. Götze, H. Auer, R. Finger, T. C. Hansen, H. Kohlmann, *Phys. B (Amsterdam, Neth.)* **2018**, *551*, 395-400
- [3] P. Önnerud, Y. Andersson, R. Tellgren, P. Nordblad, *J. Solid State Chem.* **1997**, *128*, 103-114.
- [4] G. Longworth, *Phys. Rev.* **166B**, *172*, 572-776.