Experimental report

Proposal:	6-05-952			Council: 10/2	014	
Title:	Study of the relaxati	dy of the relaxational/vibrational dynamics of glass-forming systems of pharmaceutical interest				
Research are	ea: Physics					
This proposal is	s a new proposal					
Main propos	ser: Federica MI	IGLIARDO				
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Local contac	ts: Jacques OLL	LIVIER				
Samples tre	ehalose C12H22O11, gly	vcerol C3H8O3				
Sumptes: a						
Instrument		Requested da	ays Allocated days	From	То	

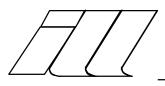
Abstract:

The carbohydrate trehalose and the alcohol glycerol, which are classified as intermediate in the Angell's strong-fragile scheme, are used as shelf-life extending additives in many labile pharmaceutical products. Since the preparation protocols used for obtaining the pharmaceutical systems affect the degree of crystallinity which plays a key role in determining the system stabilising properties and its bioprotective effectiveness, this is expected to have a significant role also in the understanding of the glycerol-trehalose mixture dynamical behaviour.

With the aim to characterize the relaxational/vibrational dynamics of the trehalose/glycerol mixture at a glycerol concentration value of 2.5% by weight, we propose to collect QENS data by the IN5B spectrometer on this mixture as obtained following two different preparation protocols, i.e. damage-formed as obtained by ball-milling and melt-formed as obtained by a rapid undercooling of the melted crystal:

- as a function of temperature across the glass transition temperature Tg at two different instrumental energy resolution values and

- as a function of energy resolution at three temperature values above, below and at Tg.



EXPERIMENT N° 6-05-952

INSTRUMENT IN5

DATES OF EXPERIMENT From : 10/07/2015 To : 15/07/2015

TITLE Study of the relaxational/vibrational dynamics of glass-forming systems of pharmaceutical interest

EXPERIMENTAL TEAM (names and affiliation)

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In the present experiment we collected QENS data by the IN5B spectrometer on the trehalose/glycerol mixture at a glycerol concentration value of 2.5% by weight.

Trehalose (alpha-D-glucopyranosyl alpha-D-glucopyranoside, $C_{12}H_{22}O_{11}$, M_w =342.3 a.u., T_g =388K) and glycerol (1,2,3-propanetriol, $C_3H_8O_3$, M_w =92 a.u., T_g =190K) are classified as "intermediate" in the Angell's strong-fragile scheme [C. A. Angell, Science 294, 2335 (2001)] and are widely employed for the preservation of biomolecules [D. Wright, A. Eroglu, M. Toner, T. L. Toth, Reprod. BioMed. 9, 179, (2004)]. Furthermore, trehalose and glycerol are also used in mixtures as shelf-life extending additives in many labile pharmaceutical products [A. Simperler et al., Phys. Chem. Chem. Phys. 9, 3999 (2007)]. However the molecular mechanisms that determine their protective effectiveness are not fully understood yet.

Previous data collected on trehalose/glycerol (T/G) mixtures in a wide concentration range (0-20%) [S. Magazù et al., J. Chem. Phys. 132, 1845121 (2010); J. Phys. Chem. B 115, 11004 (2011); J. Non-Cryst. Sol. 358, 2635 (2012)] showed an unexpected behavior for the trehalose/glycerol mixtures as a function of the glycerol content, with a reduced mean square displacement, an enhanced rigidity and a slowing down of both the fast local dynamics and the diffusive dynamics at an estimated concentration value of about 2.5% by weight of glycerol in trehalose, suggesting a correlation between the suppression of the local fast dynamics and biostability [M. T. Cicerone, C. L. Soles, Biophys. J. 86, 3836 (2004)].

We performed the experiment taking advantage from the IN5B flexibility in the wavelength selection and chopper speed in order to choose different instrumental energy resolution

values, i.e. 10, 17, 30, 50, 100 and 300 μ eV. Figure 1 shows the QENS spectra of the T/G mixture at 2.5% glycerol content at different temperature and resolution values.

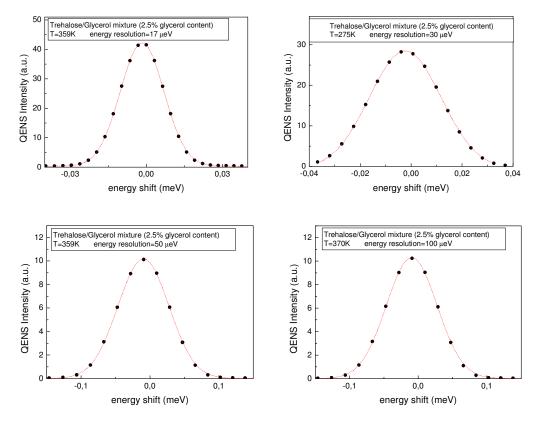


Figure 1. QENS spectra of the T/G mixture at a glycerol content of 2.5%

QENS data have been gathered at different temperature values in the range of $350\div370$ K for the energy resolution values, i.e. 10, 17, 30, 50, 100 and 300 µeV, at different temperature values in the range of $320\div390$ K at the energy resolution value of 50 µeV and at different temperature values in the range of $215\div390$ K at the energy resolution value of 30 µeV. In order to improve the data interpretation, we collected QENS data on glycerol at the temperature values of $215\cdot255\cdot295$ K at the energy resolution value of 10, 17, 30, 50, 100 and 300 µeV and on trehalose in the temperature range of $290\div359$ K at the energy resolution value of 30 µeV, at the temperature value of 350 µeV at the energy resolution value of $320\div39$ K at the energy resolution value of 10 and 17 µeV and in the temperature range of $320\div410$ K at the energy resolution value of 10 µeV. The complete analysis of the collected QENS data, with the evaluation of the intensity inflection points, i.e. the points where system relaxation time intersects the energy resolution time, is in progress. The elastic contribution should be evaluated as a function of temperature in a wide range of values in order to determine the dynamical transitions of the T/G mixture; however we were not able to register complete temperature runs due to beam-time restriction.