

Experimental report

09/06/2017

Proposal: 6-05-976

Council: 4/2016

Title: Structure of GeS₂ Glass by Neutron Diffraction with Isotope Substitution

Research area: Materials

This proposal is a resubmission of 6-05-957

Main proposer: Philip Stephen SALMON

Experimental team: Michela BUSCEMI
Anita ZEIDLER

Local contacts: Henry FISCHER

Samples: GeS₂ with ⁷⁰Ge and ⁷³Ge isotopes

Instrument	Requested days	Allocated days	From	To
D4	5	5	24/06/2016	29/06/2016

Abstract:

The method of neutron diffraction with isotope substitution will be used to measure the full set of partial structure factors for the prototypical chalcogenide glass GeS₂, which is used in the fabrication of optoelectronic and non-volatile memory devices. The results will be used to differentiate between the various models that have been proposed for GeS₂ glass, a necessary first step in identifying its structure-function relationships. The results will also be used in the search for suitable protocols for simulating the structure of GeS₂ glass by using first-principles molecular dynamics simulations, and will provide a firm basis for interpreting the structural transformations that occur to GeS₂ glass under high pressure and temperature conditions.

Atomic scale structure of GeS₂ glass

Exp. No. 6-05-976
Dates: 24/06/2016 to 28/06/2016
Experimental Team: Anita Zeidler (University of Bath)
Michela Buscemi (University of Bath)
Local Contact: Henry Fischer
Instrument: D4C

GeS₂ is one of a family of network glass-forming systems with the MX₂ stoichiometry (M = Si, Ge; X = O, S, Se) that provide a backbone matrix for materials with multiple applications. GeS₂ is used, for example, in the fabrication of optoelectronic devices on account of its high nonlinearity and photosensitivity [1-3], and as the basis of solid state electrolytes in non-volatile memory devices [4-7]. In order to understand the properties of these glasses, it is first necessary to establish the atomic-scale structure. Accordingly, there have been many experimental [8-15] and theoretical [16-20] investigations of GeS₂ glass, which has long been regarded as an important test case for examining contrasting models for topological disorder, especially as the crystal structures form both two-dimensional (2D) and three-dimensional (3D) networks under ambient conditions [21]. The structure of the glass has not, however, been established. In particular, first-principles molecular dynamics simulations of GeS₂ glass using different approaches lead to different conclusions as to the existence or not of a significant fraction of homopolar (or “wrong”) Ge-Ge and S-S bonds [16-20]. Thus, there is a need for unambiguous experimental information on the partial pair distribution functions, $g_{\alpha\beta}(r)$, for GeS₂ glass.

We decided to use the method of Ge isotope substitution to obtain unambiguous information on the partial structure factor level for GeS₂. We prepared 4 samples of GeS₂ which were identical in every respect apart from their isotopic enrichment in Ge. Specifically, we prepared ⁷⁰GeS₂, ⁷³GeS₂, ^{Nat}GeS₂, and a sample containing a 50/50 mixture of the two isotopes, ^{Mix}GeS₂. Using the D4c instrument with an incident wavelength of 0.4989(1) Å, diffraction patterns were then measured of the samples in their container, the empty container, the empty instrument, and a vanadium rod for normalisation purposes. The results will be used to inform molecular dynamics simulations.

During the experiment the polishing of the Tanzboden lead to a misalignment of the detectors, compromising the reproducibility of the background scattering and hence the accuracy of the data obtained to be able to make a full analysis. However, an example of a first order difference function where the sulphur-sulphur correlation is removed by taking the difference between $F_{70\text{GeS}_2}(q)$ and $F_{73\text{GeS}_2}(q)$ is shown in Fig. 1.

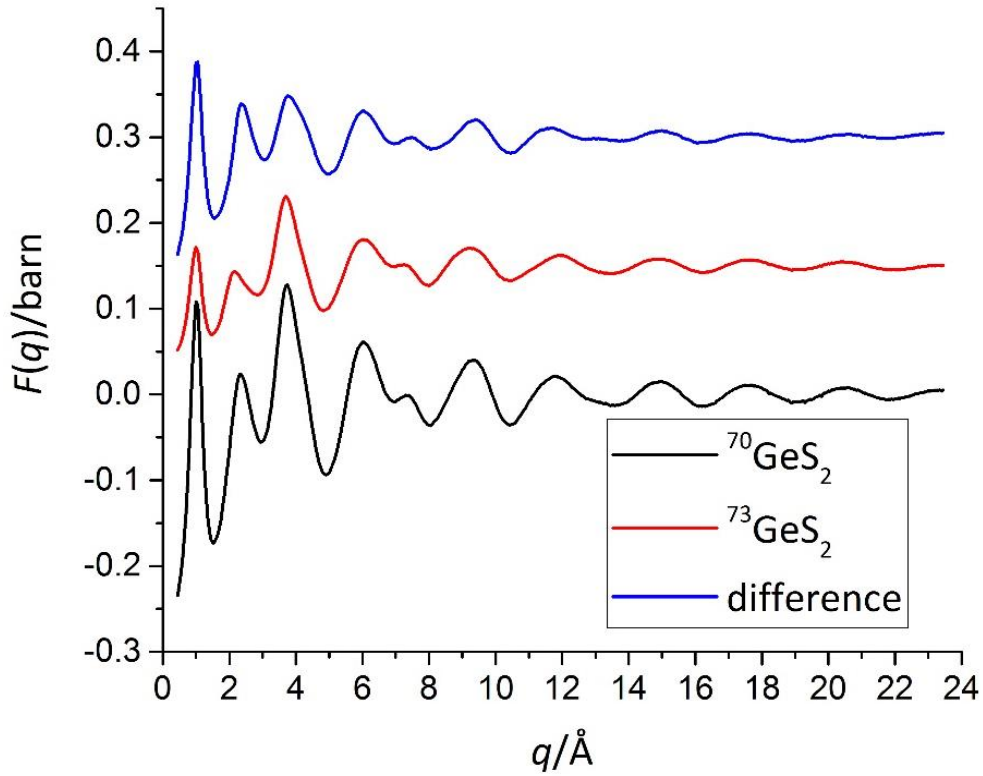


Fig. 1: The total structure factors for two of the samples and their difference function.

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