

Experimental report

17/04/2020

Proposal: 9-10-1615

Council: 10/2019

Title: Bending Moduli of CO₂ Containing Surfactants in Oil-in-Water (O/W) Microemulsions

Research area: Soft condensed matter

This proposal is a new proposal

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Samples: C₁₆H₃₃(C₂H₄O)₉(CO₂)₃

C₁₂H₂₅(C₂H₄O)₉(CO₂)₃

Instrument	Requested days	Allocated days	From	To
D33	1	1		
IN15	3	2	28/01/2020	30/01/2020
D11	0	1	02/02/2020	03/02/2020

Abstract:

Aim of this proposal is to determine the bending moduli of the monolayer of nonionic surfactants containing different amounts of CO₂ in their hydrophilic head group at the oil/water interface. This will be done by a combination of NSE and SANS on oil-in-water (O/W) microemulsions in D₂O that are saturated with d₂₂-decane in order to have a film contrast. Under these conditions combining NSE and SANS allows uniquely to determine both mean and saddle-splay modulus. This is very interesting as the CO₂ containing surfactants deviate systematically from the conventional EO surfactants with respect to their phase behaviour but also with respect to the solubilisation properties. With increasing CO₂ content in the head group the solubilisation capacity is largely enhanced. According to standard theory this should be reflected in correspondingly changed values of the bending moduli. However, so far no experimental evidence exist and NSE and SANS will allow to obtain such information much more directly and with a much higher precision, than possible with any other method. Such information will be essential to enable optimised formulations with these novel surfactants.

Bending Moduli of CO₂ Containing Surfactants in Oil-in-Water (O/W) Microemulsions

Experimental report for experiment 9-10-1615 at IN15 (Jan. 28–30, 2020)
Berlin, April 17, 2020

The aim of this proposal was to determine the bending moduli of the monolayer of nonionic surfactants containing different amounts of CO₂ in their hydrophilic head group at the oil/water interface. This has been done by a combination of NSE and SANS on oil-in-water (O/W) microemulsions in D₂O that are saturated with D₂₂-decane in order to have a film contrast. Under these conditions combining NSE and SANS allows uniquely to determine both mean and saddle-splay modulus. This is very interesting as the CO₂ containing surfactants deviate systematically from the conventional EO surfactants with respect to their phase behaviour but also with respect to the solubilisation properties. With increasing CO₂ content in the headgroup the solubilisation capacity is largely enhanced. According to standard theory this should be reflected in correspondingly changed values of the bending moduli. Such information will be essential to understand the effect of CO₂ incorporation on the properties of amphiphilic monolayers and to enable optimised formulations with these novel surfactants.

Experimental setup and samples

For this experiment we prepared microemulsions in film contrast (see Table 1). Therefore, we used deuterated decane. 50 mM of six different surfactants in D₂O were saturated with D₂₂-decane. For four samples also 80 mM geraniol was added, to extend the system to one containing a cosurfactant. Afterwards the solutions were shaken and 5 days equilibrated. The NSE data were carried out on IN15 with 5 mm cuvettes. Additionally, in-situ DLS was measured. The SANS measurements were carried out on D11 and were measured at 1.4 m and 8 m at 4.5 Å in 2 mm cuvettes. All measurements were carried out at 25°C.

Table 1: Overview of the samples measured by NSE and SANS. MAR211 and MAR212 are only measured by SANS.

Label	Long name	Surfactant	c(surfactants) \mM	c(geraniol) \mM	c(D ₂₂ -decane) \mM
MAR200	DT_C120.0_dD100	C ₁₂ (EO) ₁₄	50	-	100
MAR201	DT_C121.3_dD100	C ₁₂ (EO) _{11.4} (CO ₂) _{1.3}	50	-	100
MAR203	DT_C123.1_dD100	C ₁₂ (EO) _{8.2} (CO ₂) _{3.1}	50	-	100
MAR204	DT_C160.0_dD100	C ₁₆ (EO) _{14.6}	50	-	100
MAR205	DT_C161.5_dD100	C ₁₆ (EO) _{9.3} (CO ₂) _{1.5}	50	-	100
MAR206	DT_C162.8_dD100	C ₁₆ (EO) _{7.8} (CO ₂) _{2.8}	50	-	100
MAR207	DTOC_C120.0_G80_dD200	C ₁₂ (EO) ₁₄	50	80	200
MAR208	DTOC_C121.3_G80_dD200	C ₁₂ (EO) _{11.4} (CO ₂) _{1.3}	50	80	200
MAR209	DTOC_C123.1_G80_dD200	C ₁₂ (EO) _{8.2} (CO ₂) _{3.1}	50	80	200
MAR210	DTOC_C160.0_G80_dD200	C ₁₆ (EO) _{14.6}	50	80	200
MAR211	DTOC_C161.5_G80_dD200	C ₁₆ (EO) _{9.3} (CO ₂) _{1.5}	50	80	200
MAR212	DTOC_C162.8_G80_dD200	C ₁₆ (EO) _{7.8} (CO ₂) _{2.8}	50	80	200

Data

In Figure 1-Figure 10 the raw data is plotted as the intensity corrected by the resolution over the delay time.

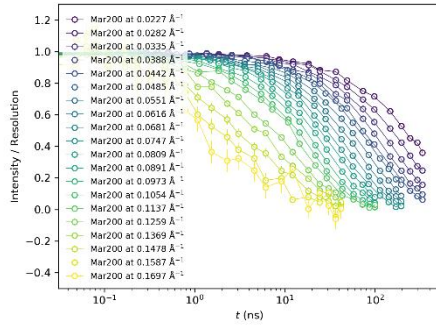


Figure 1: NSE data C_{12} -reference sample with 100 mM D_{22} -decane.

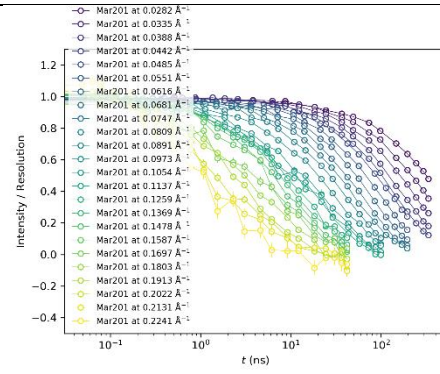


Figure 2: NSE data C_{12} -surfactant with 1.3 CO_2 -units and 100 mM D_{22} -decane.

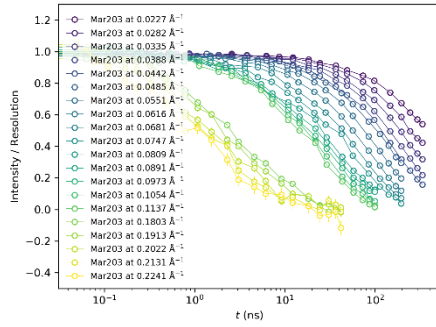


Figure 3: NSE data C_{12} -surfactant with 3.1 CO_2 -units and 100 mM D_{22} -decane.

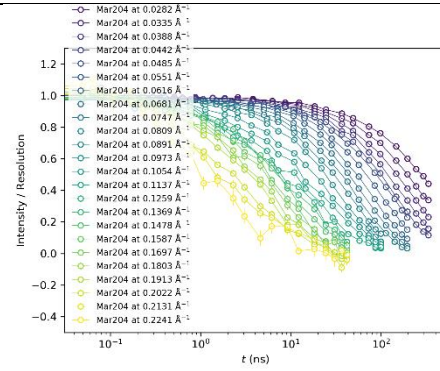


Figure 4: NSE data C_{16} -reference sample with 100 mM D_{22} -decane.

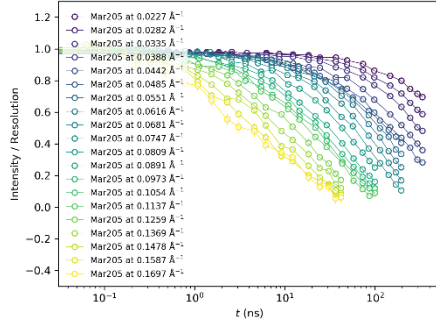


Figure 5: NSE data C_{16} -surfactant with 1.5 CO_2 -units and 100 mM D_{22} -decane.

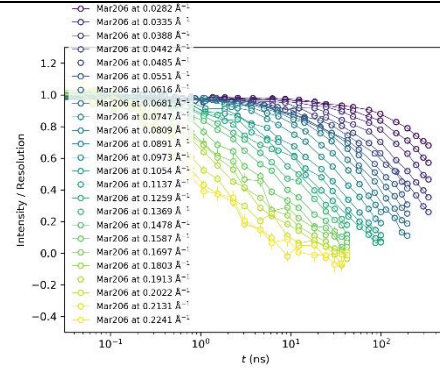


Figure 6: NSE data C_{16} -surfactant with 2.8 CO_2 -units and 100 mM D_{22} -decane.

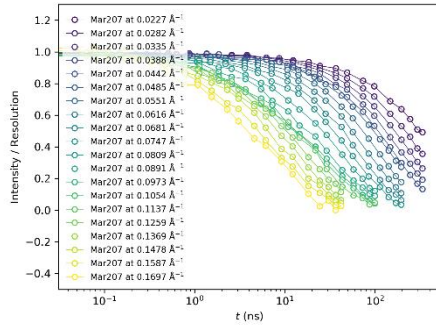


Figure 7: NSE data C_{12} -reference sample with 200 mM D_{22} -decane and 80 mM geraniol.

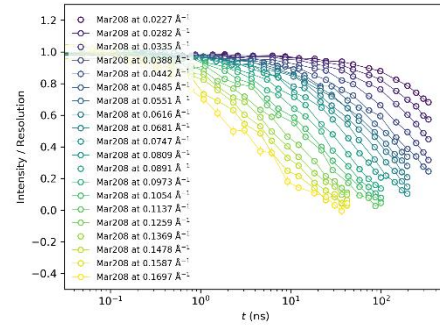
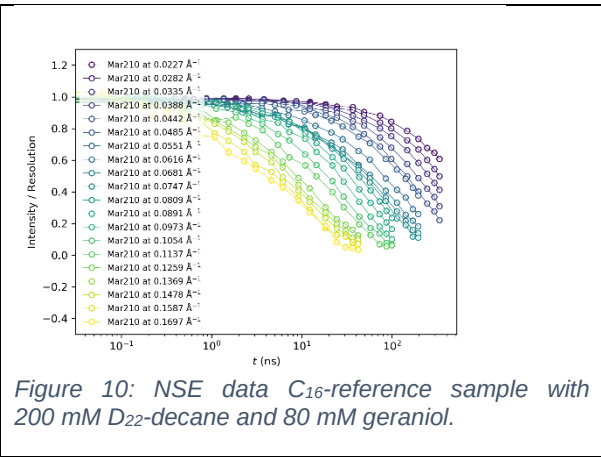
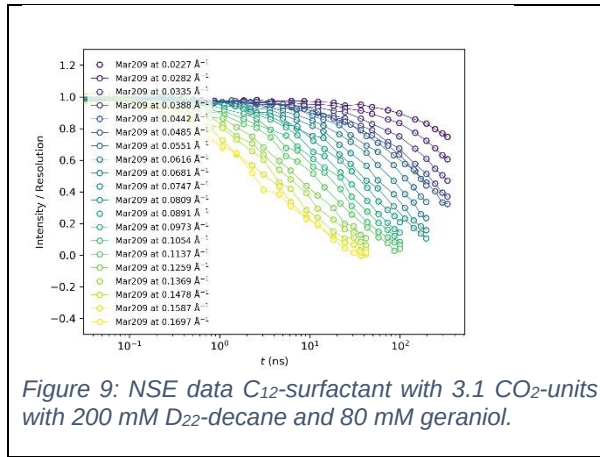
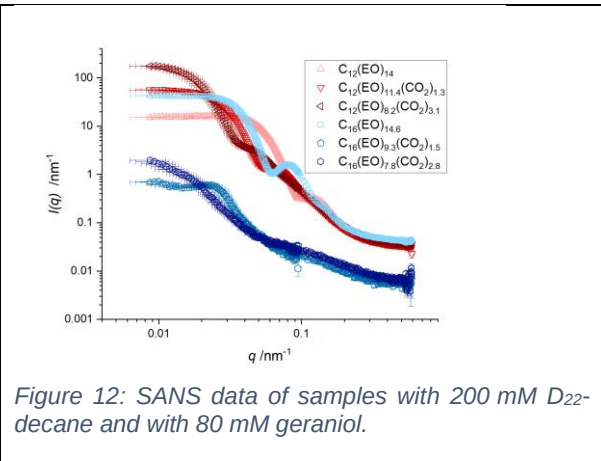
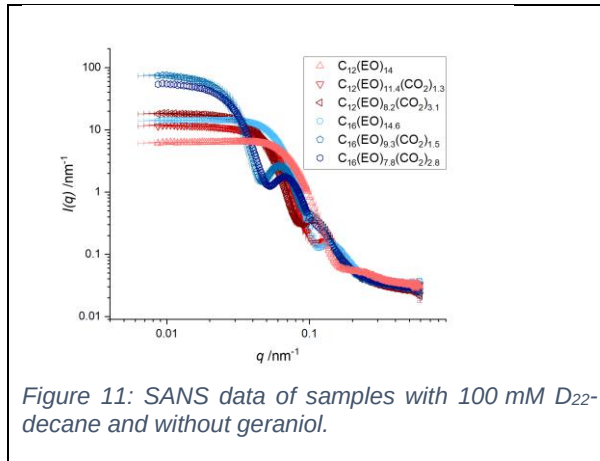


Figure 8: NSE data C_{12} -surfactant with 1.3 CO_2 -units with 200 mM D_{22} -decane and 80 mM geraniol.



The scattering data from SANS are plotted in Figure 11 and Figure 12. Figure 11 shows the samples without geraniol and Figure 12 the data with geraniol. It can be seen that the addition of oil and geraniol both leads to increasing micelle size and that those effects are increasing with increasing CO_2 content. The C_{16} -samples were hard to handle and have to be discussed carefully.



Data analysis

To evaluate the experimental data a Zilman-Granek model was used (see Figure 13), which shows poor results. In following studies the experimental data will be fitted with Milner-Safran model to improve the results and to obtain detail parameter of the bending moduli. We assume to see that the corona of the CO_2 containing systems is more flexible than the CO_2 free systems. Previous studies lead to the conclusion that the CO_2 free corona binds water stronger than the CO_2 containing.

