

Experimental report

09/02/2020

Proposal: 9-11-1967

Council: 10/2019

Title: Smart gels of sequence-controlled polymers in organic solvents

Research area: Soft condensed matter

This proposal is a new proposal

Main proposer: Francois TOURNILHAC

Experimental team: Henrique TREVISAN
Francois TOURNILHAC

Local contacts: Lionel PORCAR

Samples: Toluene D8
Toluene in toluene D8 (5 vol%)
Alt-C18 + 1/8 molar bipyridine in toluene D8 (5 vol%)
Alt-C18 + 1/4 molar bipyridine in toluene D8 (5 vol%)
Alt-C18 + 1/4 molar bipyridine D8 in toluene D8 (5 vol%)
Alt-C18 + 1/1 molar bipyridine D8 in toluene D8 (5 vol%)
Alt-C18 in toluene D8 (5 vol%)
Alt-C18 + 1/2 molar bipyridine in toluene D8 (5 vol%)
Alt-C18 + 1/1 molar bipyridine in toluene D8 (5 vol%)
Alt-C18 + 1/8 molar bipyridine D8 in toluene D8 (5 vol%)
Alt-C18 + 1/2 molar bipyridine D8 in toluene D8 (5 vol%)

Instrument	Requested days	Allocated days	From	To
D22	1	1	08/02/2020	09/02/2020

Abstract:

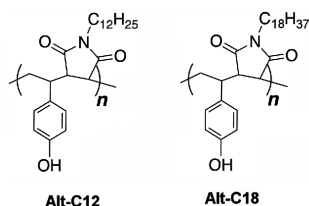
Sequence control is an emerging discipline of polymer science. In this movement, the objective of our researches is to implement novel functions in polymer materials through the control of sequence in copolymers. We previously reported physical gelation behavior of alternating copolymers consisting of monomers with long alkyl chain and phenol side groups in aromatic solvents; the SANS study performed at ILL revealed the presence of wormlike micellar objects. Since then, we observed by rheology the possibility of tuning the properties of these organogels, by controlled addition of small molecules. The current proposal aims to analyse the associated structural changes.

Smart gels of sequence-controlled polymers in organic solvents

François Tournilhac,ⁱ Henrique Trevisanⁱ

ⁱ Chimie Moléculaire, Macromoléculaire et Matériaux.
ESPCI-10 rue Vauquelin 75005 Paris, France

During this experiment beamtime, two different types of self-assembled structures made of sequence-controlled copolymers have been investigated. The base materials Alt-C12 and Alt-C18 (figure 1) are alternated copolymers of vinyl phenol and n-alkyl maleimide units, already investigated at the ILL (D22) during the DIR-158 experiment.^{1,2}



These compounds have polysurfactant structure. They show partial to complete solubility in organic solvents and insolubility in water.

In a first series of experiments, thermoreversible organogels formed in toluene at different concentrations in the presence of 1/4 molar amount of bipyridines A, B, C, D with respect to the repeat unit were investigated as a function of temperature.

In the second series of experiments, emulsions made by precipitation from solutions in a water-miscible solvent such as acetone and THF were considered. On the basis of calculated SLD values, isotope labelling was used to increase or decrease the contrast between the two solvents present in the system see description of all samples in the table in appendix.

Note: The sample named 5% alt C12 + bipy B in slot 17 in the second series is in reality 5% alt C12 + bipyD.

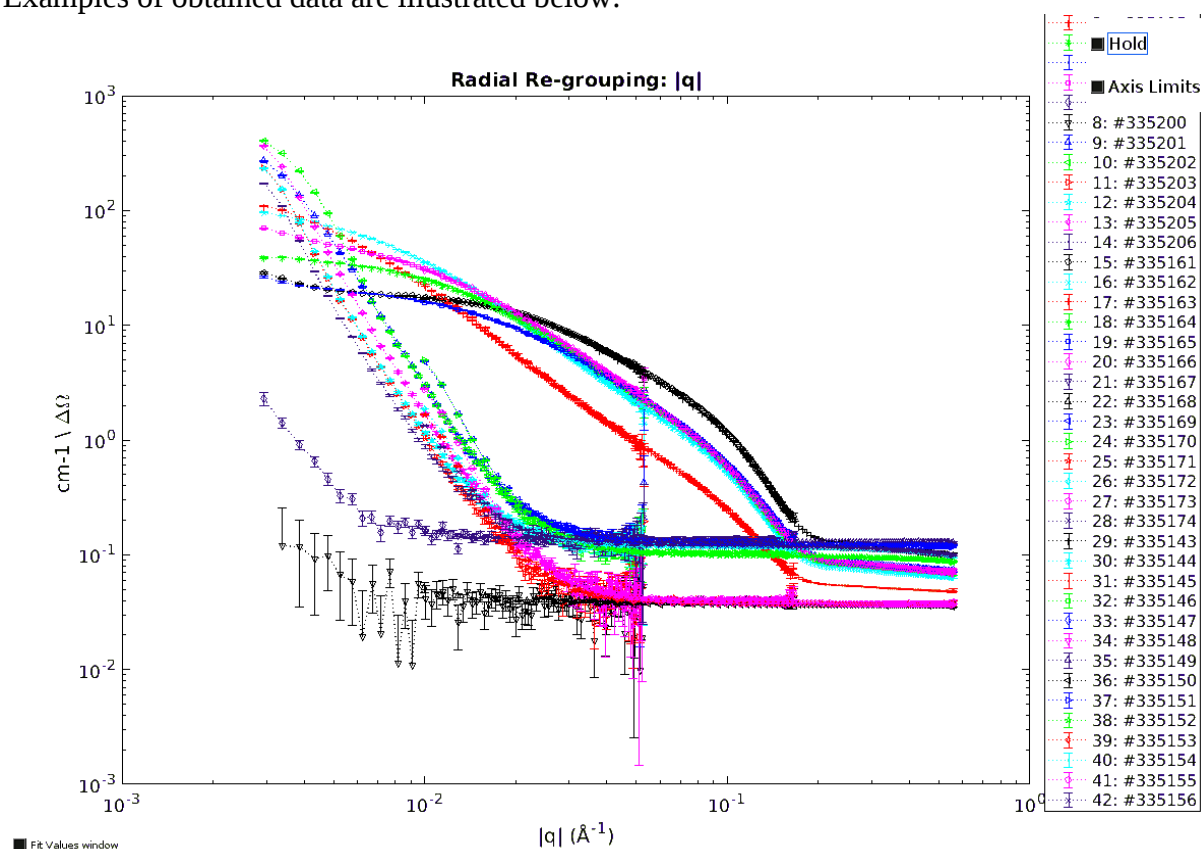
All experiments were performed on line D22 at the Institute Laue Langevin under the responsibility of Lionel Porcar. The radiation was monochromatic and sample-to-detector distances $D = 1.5$ m, 5.6 m and 17.6 m were used at $\lambda = 6$ Å. Additionally, $D = 17.6$ m was used at $\lambda = 11.5$ Å thus covering a momentum transfert range from 0.001 Å⁻¹ to 0.55 Å⁻¹. Chemistry labs at the beamline and in the science building were utilised.

Samples of solutions were prepared via dissolution in toluene D8, acetone D6, D₂O or H₂O/D₂O mixtures. Samples of emulsions were prepared by first dissolving the polymer in either acetone, THF or dioxane then adding 10-fold volume of water. All samples were analyzed at 25°C by DLS (Malvern nonosizer 90) at PSCM.

The samples were held in quartz Hellma cells with a 1 mm path; cell volume = 300 µL. The temperature was imposed by a circulating fluid in the cell's holder rack. SANS data were collected at four different temperatures, the actual temperature was measured close to the samples at any time (see table). Additionally a last scan was performed at 25°C after cooling from 90°C to ensure a data collection of all sample with an uniform temperature history.

Interestingly B10, B5 and β 10 samples (10% AltC18-bipyB, 5% AltC18-bipyB and 10% AltC12-bipyB) did clearly show a pink colour after the experiment.

Examples of obtained data are illustrated below:



Reduced data were downloaded directly from the ILL data portal site.³ SANS results in D₂O and toluene D8 will be fitted with different models with the help of SASView fitting modules to elucidate the structure of solutions and organogels. In addition for emulsions, quantitative evaluation of interphase density will be performed as described in ref 4. The results will be combined with other existing data from rheology, cryoTEM, DLS, UV-visible reflectometry and upcoming SAXS data to give a featured description of the self-assembled phases.

- 1 F. Tournilhac, E. Cazares-Cortes, M. Ouchi, L. Porcar, B. Tarus, DOI: 10.5291/ILL-DATA.DIR-158.
- 2 K. Nishimori, E. Cazares-Cortes, J.-M. Guigner, F. Tournilhac, M. Ouchi. *Polym. Chem.* 2019, **10**, 2327
- 3 F. Tournilhac, H. Trevisan, L. Porcar, DOI: 10.5291/ILL-DATA.9-11-1967.
- 4 I. Grillo, *Colloids and Surfaces A: Physicochem. Eng. Aspects* 2003, **225**, 153160

[illegible][illegible][illegible][illegible]

90 C		Tournilhac F., Trevisan H., Porcar L.					8-9 february 2020				9-11-1967	
Nb in rack	Short name	Sample name	Concentration	Solvent	SLD solid	SLD cosolvent	SLD solvent	Manip 1	Manip 2	Manip 3	Manip 4	Filename
λ (Å)								11.5	6	6	6	
D (m)								17.6	17.6	5.6	1.5	
1	MTB	Empty beam										
2	BB	Blocked beam										
3	MTC	Empty cell										
4	B10	Alt C18 + BipyB	100 mg/mL	Toluene D8	0.67		5.64	535	493	479	465	535
5	B5	Alt C18 + BipyB	50 mg/mL	Toluene D8	0.67		5.64	536	494	480	466	536
6	B2	Alt C18 + BipyB	25 mg/mL	Toluene D8	0.67		5.64	537	495	481	467	537
7	A5	Alt C18 + BipyA	50 mg/mL	Toluene D8	0.67		5.64	538	496	482	468	538
8	C5	Alt C18 + BipyC	50 mg/mL	Toluene D8	0.67		5.64	539	497	483	469	539
9	D5	Alt C18 + BipyD	50 mg/mL	Toluene D8	0.72		5.64	540	498	484	470	540
10	C18T	Alt C18	50 mg/mL	Toluene D8	0.54		5.64	541	499	485	471	541
11	C12T	Alt C12	50 mg/mL	Toluene D8	0.75		5.64	542	500	486	472	542
12	β 10	Alt C12 + BipyB	100 mg/mL	Toluene D8	0.88		5.64	543	501	487	473	543
13	β 5	Alt C12 + BipyB	50 mg/mL	Toluene D8	0.88		5.64	544	502	488	474	544
14	β 2	Alt C12 + BipyB	25 mg/mL	Toluene D8	0.88		5.64	545	503	489	475	545
15	α 5	Alt C12 + BipyA	50 mg/mL	Toluene D8	0.88		5.64	546	504	490	476	546
16	γ 5	Alt C12 + BipyC	50 mg/mL	Toluene D8	0.93		5.64	547	505	491	477	547
17	δ 5	Alt C12 + BipyD	50 mg/mL	Toluene D8	0.93		5.64	548	506	492	478	548

[illegible]